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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided.  
1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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PART I In the space provided, write the letter of the term or  
phrase that best completes each statement or best answers  
each question. \_\_\_\_ 1. According to the kinetic-molecular  
theory, particles of an ideal gas a. attract each other but do not  
collide.

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### **Chapter 9 Modern Chemistry Review Answers**

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a. S in  $\text{H}_2\text{SO}_3$  6 b. S in  $\text{MgSO}_4$  2 c. S in  $\text{K}_2\text{S}$  1 d. Cu in  $\text{Cu}_2\text{S}$  6 e. Cr in  $\text{Na}_2\text{CrO}_4$  5 f. N in  $\text{HNO}_3$  4 g. C in  $(\text{HCO}_3)_2$  ...

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